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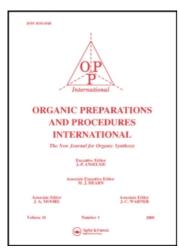
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SYNTHESIS OF SOME SPIRO α -METHYLENE- γ -LACTONES

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SYNTHESIS OF SOME SPIRO α-METHYLENE-γ-LACTONES

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A number of natural terpenoid lactones having the α -methylene- γ -butyrolactone moiety have important biological properties including antitumor activity. In connection with a search for antitumor agents, we had occasion to prepare several spiro α -methylene- γ -butyrolactones. Lactonization of the hydroxy esters resulting from the Reformatsky reaction of ethyl α -(bromomethyl) acrylate with carbonyl compounds had previously been carried out by boiling in hydrochloric acid for 4 hrs., often in quite modest yields. We now report that the lactonization may

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be promoted in 5-10 minutes by an ice-cold solution of 5% sulphuric acid from which the lactones crystallized in good yields (65-93%) and in 95% or greater purity. Ten new spiro lactones have been prepared from varying types of ketones by this technique (see table).

EXPERIMENTAL

Starting materials 6-methoxy-1-indanone, 5,6-dimethoxy-1-idanone, 2-methoxybenzosuberone, 4-chromanone and 6-chloro-4-chromanone were obtained commercially (Aldrich). 6,7-Dimethyl-3,4-dihydro-1-(2H)-naphthalenone was purchased from Chemical Samples Co. 3,4-Dihydro-5,6,7-trimethoxy-1 (2H)-naphthalenone, 3,4-dihydro-6,7,8-trimethoxy-1(2H)-naphthalenone, 6,7-dimethoxy-3,4-dihydro-1-(2H)-naphthalenone and thiachroman-4-one-1, 1-dioxide were prepared by published procedures.

General Procedure for the Preparation of the Spiro a-Methylene-y-lactones. Activated zinc metal (20 mesh; 0.8 g, 0.012 g atom, Baker Analyzed), 3,4dihydro-6,7,8-trimethoxy-1(2H)-naphthalenone (2.36 g, 0.01 mol) and dry (distilled from LAH) THF (15 ml) were placed in a dry three-necked, roundbottomed flask equipped with a magnetic stirrer, thermometer, nitrogen inlet and an additional funnel. A solution of ethyl α -(bromomethyl)acrylate (2.1 g, 0.011 mol) in dry THF (25 ml) was added dropwise during the course of 30 minutes, and the temperature of the reaction mixture was maintained at $40-45^{\circ}$. The reaction mixture turned black and a slightly exothermic reaction occurred. After being stirred at 40-45° for an additional 3 hr, the mixture was cooled and poured into ice-cold sulfuric acid 5% (200 ml). The resulting mixture was stirred until crystallization occurred (5 to 10 min), and the product was filtered and washed several times with water mp. 115-1190. An analytical sample was obtained upon recrystallization from methanol to yield colorless needles (2.12 g; 69%) of 4, mp. 122-123°. See table for additional data on the other compounds. Acknowledgment. We gratefully acknowledge support of this work by the U.S.P.H.S., National Cancer Institute, Grant CA 14343. We also express our thanks to the National Science Foundation for grants to purchase the

XL-100(15) spectrometer (Grant No. GP 17641) and its pulse-FT accessory

(Grant No. HE76-05571).

6

7

8

9

10

76

73

65

93

62

				Analys PMR ^a Calcd (I		Found) IRD	
Cmpd	Yield (%)	MP C	C=CH ₂	Allylic CH ₂	С	Н	vc=0 vc=cH ₂
1	7.2	96-97 ^d	5.69(t,1H)	3.14	73.02	6.13	1754
2	71	115-116 ^d	6.29(t,1H) 5.70(t,1H)	(q,2H) ^c 3.14	69.21	(6.24) 6.19	1664 1742
3	72	117-118 ^d	6.35(t,1H) 5.68(t,1H)	(q,2H) ^c 3.04-3.1	(68.97) 2 70.05	(6.10) 6.61	1650 1754
4	69	122-123 ^d	6.38(t, _{1H})	(m,2H)	(70.07) 67.08	(6.71)	1661
·			5.56(t,1H) 6.25(t,1H)	3.08 (q,2H) ^c	(66.98)	6.62 (6.65)	1770 1667
5	67	87-88 ^d	5.67(t,1H)	3.08	67.08	6.62	1773

6.32(t, 1H)

5.64(t,1H)

6.30(t,1H)

5.57(t,1H)

6.29(t,1H)

5.72(t,1H)

6.35(t, 1H)

5.77(t,1H)

6.37(t, 1H)

5.88(t, 1H)

6.22(t,1H)

3.10

3.14

3.16

3.15

2.82

(t,2H)

(t,2H)

(q,2H)^C

(q,2H)^C

(q,2H)^C

(q,2H)^C

(66.70)

79.30

(79.18)

74.39

(74.28)

72.20

(71.96)

62.28

(62.17)

59.09

(59.03)

(6.47)

7.49

(7.41)

7.02

(6.99)

5.59

(5.70)

4.42

(4.58)

4.57

(4.47)

1639

1770

1658

1761

1664

1770

1667

1760

1658

1770

1653

Table I. Physical and Spectral Data of Spiro α-Methylene-γ-lactones

REFERENCES

+ Research Associate, 1975-present

73-74^d

104-105^d

156-157[‡]

- 1. S.M. Kupchan, R. J. Hemingway, D. Werner, A. Karim, A. T. McPhail, G. A. Sim, J. Am. Chem. Soc., 90 3596 (1968); S.M. Kupchan, R.J. Hemingway, J. M. Cassady, R. J. Knox, A. T. McPhail, G. A. Sim, ibid., 89, 465 (1967); S. M. Kupchan, Y. Aynehchi, J. M. Cassady, A. T. McPhail, G. A. Sim, H. K. Schoes, A. L. Burlingame, ibid., 88 3674 (1966). S. M. Kupchan, R. J. Hemingway, D. Werner, A. Karim, J. Org. Chem., 34, 3903 (1969); S. M. Kupchan, J. E. Kelsey, M. Maruyama, J. M. Cassady, R. J. Hemingway, J. R. Knox, ibid; 34, 3876 (1969); S. M. Kupchan, J. E. Kelsey, M. Maruyama, J. M. Cassady, Tetrahedron Lett, 3517 (1968).
- A. Loffler, R. D. Pratt, J. Pucknat, G. Geibard, A. S. Dreiding, Chemica, 23, 413 (1969); E. Ohler, K. Reininger, U. Schmidt, Angew. Chem., Internat. Ed. Eng., 9, 457 (1970). A general review for the preparations of these lactones has been released recently; see

PMR recorded in CDCl3 on Varian XL-100(15) spectrometer, expressed in δ values relative to TMS. b. Recorded on a Beckman IR-5A Infrared spectrometer as KBr pellets and expressed in cm -1. c. Complex quartet. The remaining spectral features were entirely consistent with the structures and full spectra can be obtained upon request. d. Crystallized from methanol. e. Crystallized from ethanol. Crystallized from 2-propanol.

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- P. A. Grieco, Synthesis, 67 (1975); A. Rosowsky, N. Papathana Sopoulos, H. Lazarus, G. E. Foley, E. J. Modest, J. Med. Chem., <u>17</u>, 672 (1974); K. H. Lee, T. Ibuka, S. H. Kim, B. R. Vestal, I. H. Hall, and E. S. Huang, J. Med. Chem., <u>18</u> 812 (1975).
- E. A. Mawdsley, K. D. Berlin, R. W. Chesnut and N. N. Durham, J. Med. Chem., 19 239 (1976).
- M. M. Hashem, K. D. Berlin, R. W. Chesnut and N. N. Durham, ibid., <u>19</u>, 229 (1976).
- 5. I. W. J. Still and M. T. Thomas, J. Org. Chem., 33, 2730 (1968).
- 6. M. S. Newman and F. J. Evans, Jr., J. Am. Chem., 77, 946 (1955).
- 7. A. F. Ferris, J. Org. Chem., 20, 780 (1955).

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